

**Open Access** 

**Research Article** 

# Study of Groundwater Quality using GIS in Hirehalla Sub-basin of Koppal District, Karnataka, India

Sreenivasa A. and Ajay Kumar N. Asode

Department of Studies in Geology, Karnatak University, Dharwad, Karnataka, India

Publication Date: 12 April 2016

Article Link: http://scientific.cloud-journals.com/index.php/IJAESE/article/view/Sci-419



Copyright © 2016 Sreenivasa A. and Ajay Kumar N. Asode. This is an open access article distributed under the **Creative Commons Attribution License**, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

**Abstract** Water quality problems were reported in the groundwaters of Hirehalla Sub-basin, Koppal district, Karnataka, India. 21 representative samples were collected and subjected to hydrochemical investigations. Major cations- Ca, Mg, Na, K and anions- CO<sub>3</sub>, HCO<sub>3</sub>, Cl, SO<sub>4</sub> were analysed. Physical parameters such as pH, EC, TDS, H were also determined. Piper hill trilinear diagram were plotted to study the hydrochemical characters of groundwater and it was found the groundwater to be of class- NaCl and CaMgCl type. Fluoride ion concentration was found to be in the range 0.26 to 6.42 mg/lit. Based on EC values it was found that 61% of water samples fall in excellent category. Drinking use suitability was determined by comparing the analysis results of water samples with BIS and WHO standards and it was found that 65% of the water samples are safe and fit for drinking purposes. Parameters like SAR and Na% were calculated to evaluate the suitability of waters for irrigational purposes. USSL and Wilcox diagram were plotted to determine agriculture suitability. It was found that 90% of the water samples are fit for irrigation. Different spatial variation maps were generated in GIS environs. Thus GIS based approach was efficient tool in assessment of water quality of Hirehalla Sub-basin.

Keywords Hirehalla Sub-Basin; Hydrogeochemistry; Water Quality; Fluoride; SAR; GIS

#### 1. Introduction

The quality of water is one of the major concern and problem of almost all nations of the world. Not all of the available water resources on the globe can be used directly for drinking, agriculture and industrial purposes. While of the available limited direct resources many have the problem of water quality some are due to natural processes (geogenic) and others due to anthropogenic activities. Thus making mankind to explore other source of water such as groundwater to fulfil and meet his demand for different purposes.

In this quest of finding water resources, suitable groundwater potential sites are being located and explored. Due to this over-exploitation the water table levels are slowly lingering deeper and deeper. And with the irregular trend of monsoons and no other source of recharge the groundwaters are depleting at faster rates.

The quality of water resources is governed by many factors such as climate, topography, soil characters, sub-surface litho-units, groundwater flow, anthropogenic activities (Rajesh et al., 2002; Lakshmanan et al., 2003; Srivastava, 2005; Das Brijraj and Kaur, 2007; Cloutier et al., 2008; Prasanna et al., 2010). The study area in particular being a typical hard rock terrain of Peninsular India consisting predominantly granites and gneiss here it is mainly controlled by factors such as weathering and its features and impact of anthropogenic activities.

Many researchers and scientists from different parts of the world have studied the water quality in different geologic regimes with both conventional methods and GIS approach (Sreedevi, 2002; Anbazhagan and Archana, 2003; Subbarao and John Devdas, 2003; Subbarao, 2006; Subramani et al., 2010).

Thus in the present study an attempt has been made to assess the water quality of Hirehalla Subbasin with the help of GIS. Further suggest the respective authorities for proper planning, management and development of groundwater resources.

# 2. Study Area

The present study area Hirehalla Sub-basin of Koppal district, Karnataka, India is geographically located between  $15^{\circ}23$ ' to  $15^{\circ}32$ ' N latitude and  $75^{\circ}45$ ' to  $76^{\circ}07$ ' E longitude and is covered in Survey of India (SOI) toposheets numbered 48M/14, 48M/15 and 57A/03 and spread over an area of 389 km<sup>2</sup>. It is a tributary of Tungabhadra River.

Physiographically major portion is covered by plains with shallow troughs and mounds at some places. The highest and lowest elevations are 625 and 552 meters respectively. The major rock types include granitic gneiss and schist. These rocks are highly weathered and the resulting fractures, joints act as conduit for storage of water. The waters are found to be in phreatic conditions. The study area experiences semi-arid type of climate which is characterized by hot summer where maximum temperature reaches upto  $45^{\circ}$ C in the month of April and May while coolest month is December and January where temperature drops upto  $16^{\circ}$ C (Figure 1).



Figure 1: Location Map of Hirehalla Sub-basin

#### 3. Materials and Methods

Water samples were collected in one litre polyethylene cans which were pre-rinsed and washed in the laboratory using proportionate distilled water and dil. HCl and later in the field it was washed using representative sample. Standard procedure of analysis was followed that of APHA (APHA, 1995). Table 1 depicts the list of parameters the samples were subjected for analysis and the methodology/instrument applied for same. Table 2 shows the results obtained after physico-chemical analyses. All the thematic and variation maps were generated in GIS environ using different tools of arcGIS v10 such as spatial analyst, interpolation.

SI. No.	Parameters	Analytical Methods/Instruments
1	рН	Elico LI120
2	Electrical Conductivity (EC)	Elico CM180
3	Total Dissolved Solids (TDS)	EC*0.65 (Todd 1980)
4	Hardness (H)	Trimetric
5	Calcium (Ca)	Trimetric
6	Magnesium (Mg)	Calculation
7	Sodium (Na)	Atomic Absorption Spectrophotometer (AAS)
8	Potassium (K)	Atomic Absorption Spectrophotometer (AAS)
9	Carbonate (CO <sub>3</sub> )	Trimetric
10	Bicarbonate (HCO <sub>3</sub> )	Trimetric
11	Chloride (Cl)	Trimetric
12	Sulphate (SO <sub>4</sub> )	Colorimeter
13	Fluoride (F)	Ion Selective Method

#### Table 1: List of Physico-Chemical Parameters and Analytical Methods

Table 2: Physico-Chemical Analyses Results

SI No	LOCATION	TDS	EC	Н	pН	Ca	Mg	Na	K	CO3	HCO3	đ	SO4	F
1	HIREHANCHIHALLA	320	490.61	120	79	24.04	14.61	109.53	2.19	0	180	63.9	60	1.1
2	BINNALA	2860	443.41	1752	8.11	352.7	212.46	415.72	1.86	0	250	1356.1	710	0.69
3	CHIKENAKOPPA	3030	460.97	2128	7.58	432.86	255.34	494.96	1.77	0	225	2157.3	380	0.26
4	SHIDNEKOPPA	1920	297.67	1260	75	192.38	190.04	476.21	15	0	365	1262.6	700	0.43
5	SOMPUR	7840	124.8	3220	7.16	838.47	274.83	<b>84</b> 5	2.38	0	350	3494	380	0.35
6	MALEKOPPA	5370	102.32	2640	7.28	585.16	287.5	820	1.94	0	325	3061.2	380	0.36
7	MANNAPUR	2090	324.03	1060	7.49	157.11	162.75	656.7	7.87	0	375	1704.22	560	1.98
8	LINGAPUR	1620	251.16	680	7.61	128.25	87.71	421.15	751	0	280	681.6	550	2.29
9	LINGAPUR	1180	182.94	404	7.93	64.12	59.44	392.54	6.5	0	320	731.3	430	2.33
10	ITAGI	1100	170.54	720	7.64	128.25	97.45	179.57	7.88	0	300	546.7	450	0.44
11	MANDALGERI	1380	213.95	1260	7.55	264.52	146.18	226.18	9.25	0	205	1072.1	600	0.51
12	BETAPANAHALLI	5570	863.56	2440	7.6	689.37	175.42	746	9.01	0	355	2350	380	0.47
13	GORLEKOPPA	620	960.12	320	1.71	40.08	53.6	196.04	5.31	0	360	79.52	170	1.89
14	KUKNUR	1150	178.29	940	7.29	272.54	63.34	134.39	5.2	0	275	809.4	268	0.44
15	GAVARHALLA	600	930.02	300	7.83	64.12	64.11	155.92	2.36	0	265	191.7	70	0.59
16	CHENDURU	1390	215.5	960	7.32	173.14	128.64	301.37	5.15	0	460	898.86	275	1.08
17	VIRAPUR	970	150.38	212	7.83	48.09	22.41	383.66	2	0	525	156.2	312	6.42
18	MUDLAPUR	370	570.36	100	8.2	19.23	12.66	151.2	1.59	0	210	107.92	80	2.02
19	BENAKAL	2100	325.58	200	8.28	40.08	24.36	651.03	2.21	0	<i>T1</i> 5	866.2	350	5.94
20	NITALLI	2350	364.34	1760	7.27	288.57	253.39	434.02	6.21	0	335	1830	600	1.15
21	MOSOBAHANCHINALA	1460	226.35	980	73	296.59	58.47	269.3	4.42	0	350	781	400	0.57

#### 4. Results and Discussion

# 4.1. Water Quality

### 4.1.1. Electrical Conductivity (EC)

EC value in the study area range from 102 to 960  $\mu$ S/cm. Water samples are classified based on Sarma et al. (1982) (Table 3) and accordingly it was found that approximately 62% of water samples fall in excellent category followed by 19% each falling in good and permissible category. Figure 2 shows the EC distribution in the study area.

Class	EC range in µS/cm	No. of Samples	Percentage
Excellent	0-333	13	61.90
Good	333-500	04	19.04
Permissible	500-1000	04	19.04
Brackish	1000-1500	-	-
Saline	1500-10,000	-	-

Table 3: EC Classification based on Sarma et al.

### 4.1.2. Total Dissolved Solids (TDS)

TDS can be defined as the different types of mineral present in water in the dissolved state. To ascertain the suitability of groundwater for any purpose, TDS is an important factor. Thus the waters of the study area were classified based on Freeze and Cherry (1979) (Table 4). It suggests that 76% of the water samples in the Hirehalla Sub-basin are brackish while 24% are fresh waters. Figure 3 shows the distribution of TDS in the study area.

Fable 4: Classification of	TDS based on	Freeze and Cherry
----------------------------	--------------	-------------------

TDS (mg/lit)	Nature of Water	No. of Samples	%
<1000	Fresh	05	23.80
1000-10000	Brackish	16	76.19
10000-100000	Saline	-	-
>100000	Brine	-	-

# 4.1.3. Hardness (H)

The classification of water samples of Hirehalla Sub-basin based on Sawyer and McCarthy (1967) (Table 5) shows that 85% of the water samples fall in water class hard while 9.52% fall in moderately hard and 4.76% fall in very hard class. Figure 4 shows the distribution of hardness in the study area.

Table 5: Classification of Waters Based on Hardness

Hardness (CaCO <sub>3</sub> ) in ppm	Water Class	No. of Samples	%
0-75	Soft	-	-
75-150	Moderately Hard	02	9.52
150-3000	Hard	18	85.71
>3000	Very Hard	01	4.76



Figure 2: Spatial distribution of Electrical Conductivity (EC)



Figure 3: Spatial distribution of Total Dissolved Solids (TDS)



Figure 4: Spatial distribution of Hardness (H)

# 4.2. Hydrochemical Facies (Piper Hill Tri-linear Diagram)

The chemical processes and the evolution of the groundwater in the aquifer due to prolonged presence and flow may be assessed using the hydrochemical facies. This can be best illustrated and interpreted using Hill Piper plot wherein plot of cations mainly Ca, Mg, Na+K versus anions  $HCO_3$ , Cl,  $SO_4$  is plotted. From the plot (Figure 5) it was found that the groundwaters of Hirehalla Sub-basin falls

in field 2 & 4 and are categorized as -NaCl and Mixed CaMgCl type. This indicates strong acid exceeding weak acid and alkali exceeding alkaline.



Figure 5: Piper's Diagram for the waters of Hirehalla Sub-basin

### 4.3. Drinking Quality

Physico-chemical analyses results of the Hirehalla Sub-basin were compared with Bureau of Indian Standards (BIS, 2003) and World Health Organisation (WHO, 2005) to decipher the quality of water for drinking purpose (Table 7). The comparison study revealed that on an average 65% of the water samples for all parameters except TDS fall in allowable limits and is safe for drinking purpose while 76% of the water samples exceed allowable limit based on TDS. Thus there is need to check the water quality and necessitate appropriate actions for supply of safe drinking water.

	Drinking Water Standards				Samples	
Parameters	BIS (2003)		WHO (2	005)	exceeding	%
	Permissible	Allowable	Permissible	Allowable	allowable limit	
рН	7.0-8.5	6.5-9.2	7-8.5	6.5-9.2	-	-
TDS	NS	NS	500	1000	16	76.19
TH	300	600	-	1000	09	42.85
Ca	75	200	75	200	08	38.09
Mg	50	150	50	150	08	38.09
CI	250	1000	200	600	09	42.85
SO <sub>4</sub>	200	400	200	400	08	38.09
F	1.0	1.50	-	1.50	07	33.33

Table 7: Drinking Water Standards of BIS (2003) and WHO (2005)

#### 4.4. Irrigational Quality

# 4.4.1. Sodium Absorption Ratio (SAR)

SAR is considered a better measure of sodium/alkali hazard in irrigation as it is directly related to the adsorption of sodium by soil and is an important criterion in determining the suitability of water for irrigation. The SAR is classified as shown in Table 8 and is calculated using:

$$SAR = \frac{Na^{+}}{(\sqrt{Ca^{2+} + Mg^{2+}})/2}$$
 all values are in epm

According to classification (Table 8), it shows that 90% of water samples in the study area fall in excellent category while 5% each fall in good and fair category. Groundwater samples were also plotted in United States Salinity Laboratory diagram (Richard, 1954) to study its suitability for irrigation where plot of SAR vs EC is plotted (Figure 6) and it was found that, 42.85% of water samples fall in field of C1S1, indicating low salinity and low sodium, 23.80% of water samples fall in field of C2S1, indicating medium salinity and low sodium, 14.28% samples falling in field of C3S1, indicating high salinity and low sodium and 9.52% samples falling in field of C2S2, indicating medium salinity and medium sodium. Figure 8 shows the spatial distribution of SAR.

Class	SAR Range	No. of Samples	Percentage
Excellent	<10	19	90.47
Good	10-18	01	4.76
Fair	18-26	01	4.76
Poor	>26	-	-

Table 8: Classification based on SAR



Figure 6: USSL Diagram for Classification of Irrigation Waters, 1950

#### 5.4.2. Sodium Percent (Na%)

The suitability of the groundwater for irrigation depends on the minerals and its interaction present in water and its effect on plants and soil. When the concentration of sodium is high in irrigation water, sodium ions tend to be absorbed by clay particles which reduce soil permeability (Saleh, 1999; Arumugam and Elangovan, 2008). This can be computed using:

$$\%Na = \frac{(Na^{+} + K^{+}) \times 100}{(Ca^{2+} + Mg^{2+} + Na^{+} + K^{+})}$$
 all values are in epm

The classification for irrigation suitability based on sodium percent (Table 9) revealed that 66.66% of water samples from the study area fall in category excellent while 33.33% fall in good category. Similar comparison study to know the suitability of water for irrigation is done by plotting the plot of sodium percent vs EC i.e., Wilcox diagram (Wilcox, 1954) (Figure 7) which indicate 85% of the water samples falling in range excellent to good category and 15% of water samples fall in good to permissible category. Figure 9 shows the spatial distribution of sodium percent.

Table 9: Sodium Percent Classification for Irrigation Water

Class	Na%	No. of Samples	Percentage
Excellent	< 20%	14	66.66
Good	20-40	07	33.33
Permissible	40-60	-	-
Doubtful	60-80	-	-
Unsuitable	>80	-	-



Figure 7: Classification of Irrigated Waters based on Wilcox, 1955



Figure 8: Spatial Distribution of Sodium Absorption Ratio (SAR)



Figure 9: Spatial distribution of Sodium percent

# 6. Conclusion

GIS based approach has turned out to be very effective and efficient tool in determining the water quality of hard rock Hirehalla Sub-basin, Koppal district, Karnataka were water quality issues has been reported. Physico-chemical analyses result obtained revealed that, based on TDS 76% of the water samples are brackish and 85% hard based on hardness. Comparison of water samples with BIS and WHO standards for suitability for drinking purpose indicated approximately 35% of the water samples having water quality problem with respect to domestic use.

SAR classification for irrigational suitability showed 90% of the water samples in excellent category. Plot of SAR vs EC (USSL diagram) indicated the water samples to be in the field C1S1, C2S1 & C3S1. Based on sodium percent classification for irrigated waters it was found that 66.66% of water samples range in water class excellent and 33.33% in good. Wilcox diagram revealed the water samples to be ranging from excellent to permissible. Fluoride incidence was reported at some places which are due to geogenic processes. Further it is suggested to the authorities to keep a check on the water quality at regular intervals and necessitate actions for providing safe drinking water. Furthermore plan and manage for sustainable development of water resources.

#### Acknowledgement

The authors thank indepthly Dr. H.M. Jayasheela, Chairman, Dept. of Studies in Geology for providing all the necessary facilities to carry out this work. Thanks are due to the anonymous reviewers for their suggestions to uplift the quality of paper.

#### References

American Public Health Association (APHA) 2005: *Standard Method for Examination of Water and Wastewater.* 21st Edn. APHA, AWWA, WPCF, Washington DC, USA.

Arumugam, K. and Elangovan, K. *Hydrochemical Characteristics and Groundwater Quality Assessment in Tirupur Region, Coimbatore District, Tamil Nadu, India.* Environ Geol. 2008. 58; 1509-1520.

BIS, 2003: Drinking Water Specifications. Bureau of Indian Standards IS: 10500.

Cloutier Vincent, Lefebvre Rene, Therrien Rene and Savard Martine M. *Multivariate Statistical Analysis of Geochemical Data as Indicative of the Hydrogeochemical Evolution of Groundwater in a Sedimentary Rock Aquifer System.* Journal of Hydrology. 2008. 353; 294-313.

Das Brijraj, K. and Kaur, P. *Geochemistry of Surface and Subsurface Waters of Rewalsar Lake, Mandi District, Himachal Pradesh: Constraints on Weathering and Erosion.* Journal Geological Society of India. 2007. 69 (5) 1020-1030.

Eaton, EM. Significance of Carbonate in Irrigation Water. Soil Sci. 1950. 69; 123-133.

Lakshmanan, E., Kannan, R., and Senthil Kumar, M. *Major Ion Chemistry and Identification of Hydrogeochemical Processes of Groundwater in a Part of Kancheepuram District, Tamilnadu, India.* Environmental Geosciences. 2003. 10 (4) 157-166.

Prasanna, M.V., Chidambaram, S. and Srinivasamoorthy K. *Statistical Analysis of the Hydrogeochemical Evolution of Groundwater in Hard and Sedimentary Aquifers System of Gadilam River Basin, South India.* Jour of King Saud University (Science). 2010. 22; 133-145.

Ravikumar, P. and Somashekhar, R.K. *Multivariate Analysis to Evaluate Geochemistry of Groundwater in Varahi River Basin of Udupi in Karnataka (India).* The Ecoscan. 2010. 4 (2&3) 153-162.

Piper, A.M. A Geographic Procedure in the Geochemical Interpretation of Water Analysis. Trans. Am. Geophysics Union. 1994. 25; 914-928.

International Journal of Advanced Earth Science and Engineering

Ragunath, H.M., 1987: Groundwater. Wiley Eastern, New Delhi. 563.

Rajesh Reghunath, Sreedhara Murthy, T.R., and Raghavan, B.R. *The Utility of Multivariate Statistical Techniques in Hydrogeochemical Studies: An Example from Karnataka, India.* Water Research. 2002. 36; 2437-2442.

Richard, L.A., 1954: *Diagnosis and Improvement of Saline Alkali Soils.* US Department of Agriculture, Hand Book. 60; 160.

Anbazhagan S. and Archana M. Nair. *Geographic Information System and Groundwater Quality Mapping in Panvel Basin, Maharashtra, India.* Environmental Geology. 2004. 45 (6) 753-761.

Saleh, A., Al-Ruwaih, F., and Shehata, M. *Hydrogeochemical Processes Operating within the Main Aquifers of Kuwait.* J Arid Environ. 1999. 42; 195-209.

Sarma, V.V.J., Prasad, N.V.B.S.S., and Rajendra Prasad, P. *The Geohydrochemistry of Groundwater along Visakhapanam-Bhimilipatnam Coast with Regard to their Utility to Drinking Domestic and Irrigation Purposes.* Jour. Aso. Explo. Geophy. 1982. 2; 51-63.

Sreedevi, P.D., 2002: A Case Study on Changes in Quality of Groundwater with Seasonal Fluctuations of Pageru River Basin, Cuddapah District, Andhra Pradesh, India.

Srinivasa Rao, Y., Reddy, T.V.K., and Nayudu, P.T. *Groundwater Quality in the Niva River Basin, Chittoor District, Andhra Pradesh, India.* Environmental Geology. 1997. 32 (1) 56-63.

Srivastava Ajay. Aquifer Geometry, Basement-Topography and Groundwater Quality around Ken Graben, India. Journal of Spatial Hydrology. 2005. 2 (2) 1-7.

Subba Rao, N. and John Devadas, D. *Fluoride Incidence in Groundwater in an Area of Peninsular India.* Environ. Geol. 2003. 45; 243-251.

Subba Rao, N. Seasonal Variation of Groundwater Quality in a Part of Guntur District, Andhra Pradesh, India. Environ Geol. 2006. 49; 413-429.

Subramani, T., Rajmohan, N., and Elango, L. *Groundwater Geochemistry and Identification of Hydrogeochemical Processes in a Hard Rock Region, Southern India.* Environmental Monitoring and Assessment. 2010. 162; 123-137.

WHO, 2005: International Standards for Drinking Water, 3rd ed., Geneva. 2005.

Wilcox, L.V., 1955: *Classification and Use of Irrigation Waters.* USDA, Circular 969, Washington, DC, USA.